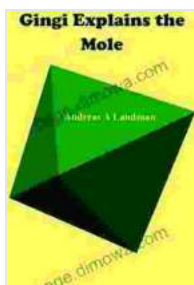


# Gingi Explains the Mole: Elementary Science

## What is the Mole?

The mole is a unit of measurement that is used to measure the amount of a substance. It is defined as the amount of a substance that contains exactly  $6.022 \times 10^{23}$  particles of that substance. This number is known as Avogadro's number.



## Gingi Explains the Mole (Elementary Science Book 2)

by Peter C. Stone

★★★★★ 5 out of 5

Language : English  
File size : 816 KB  
Text-to-Speech : Enabled  
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The mole is a very large unit of measurement. For example, one mole of water contains  $6.022 \times 10^{23}$  molecules of water. This is a lot of molecules!

## Why is the Mole Important?

The mole is an important unit of measurement because it allows us to compare the amounts of different substances. For example, we can use the

mole to compare the amount of water in a glass of water to the amount of water in a swimming pool.

The mole is also important because it is used in chemical reactions. A chemical reaction is a process in which one or more substances are transformed into one or more new substances. The mole is used to measure the amounts of the reactants and products in a chemical reaction.

## **How to Use the Mole**

The mole is a simple unit of measurement to use. To use the mole, you simply need to know the mass of the substance you are measuring and the molar mass of the substance.

The molar mass of a substance is the mass of one mole of that substance. The molar mass is expressed in grams per mole (g/mol).

Once you know the mass of the substance you are measuring and the molar mass of the substance, you can use the following formula to calculate the number of moles of the substance:

Number of moles = Mass of substance (g) / Molar mass of substance (g/mol)

## **Example**

Let's say you want to calculate the number of moles of water in a glass of water. The mass of the water in the glass is 250 g. The molar mass of water is 18 g/mol.

Number of moles = 250 g / 18 g/mol = 13.89 moles

Therefore, there are 13.89 moles of water in the glass of water.

The mole is a fundamental unit of measurement in science. It is used to measure the amount of a substance, and it is essential for understanding chemical reactions. The mole is a simple unit of measurement to use, and it is important for students to learn how to use it.

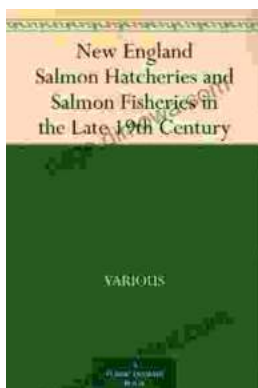


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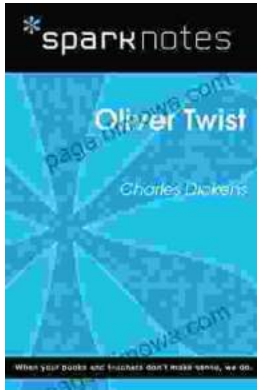
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